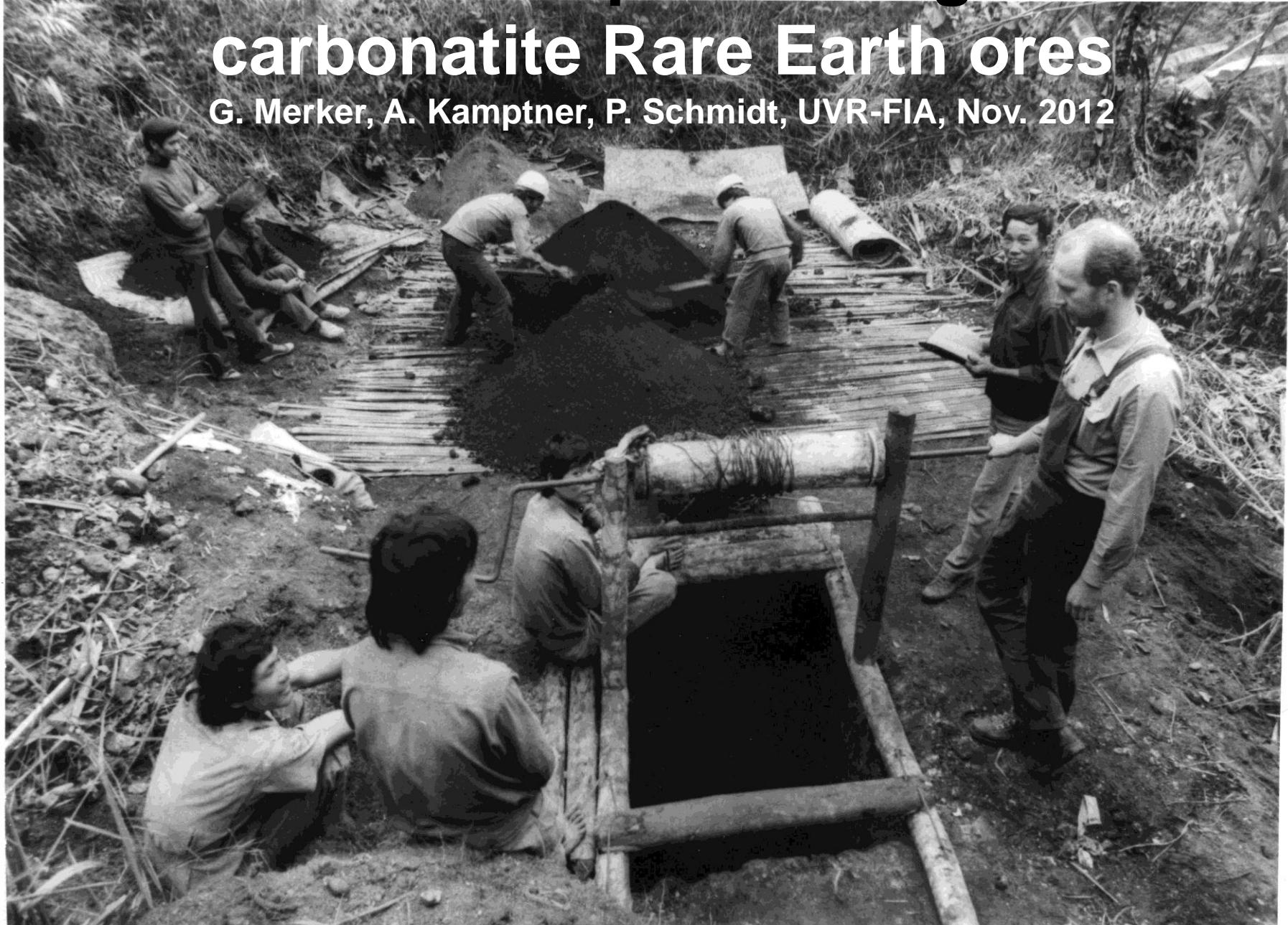


# About the processing of

# carbonatite Rare Earth ores

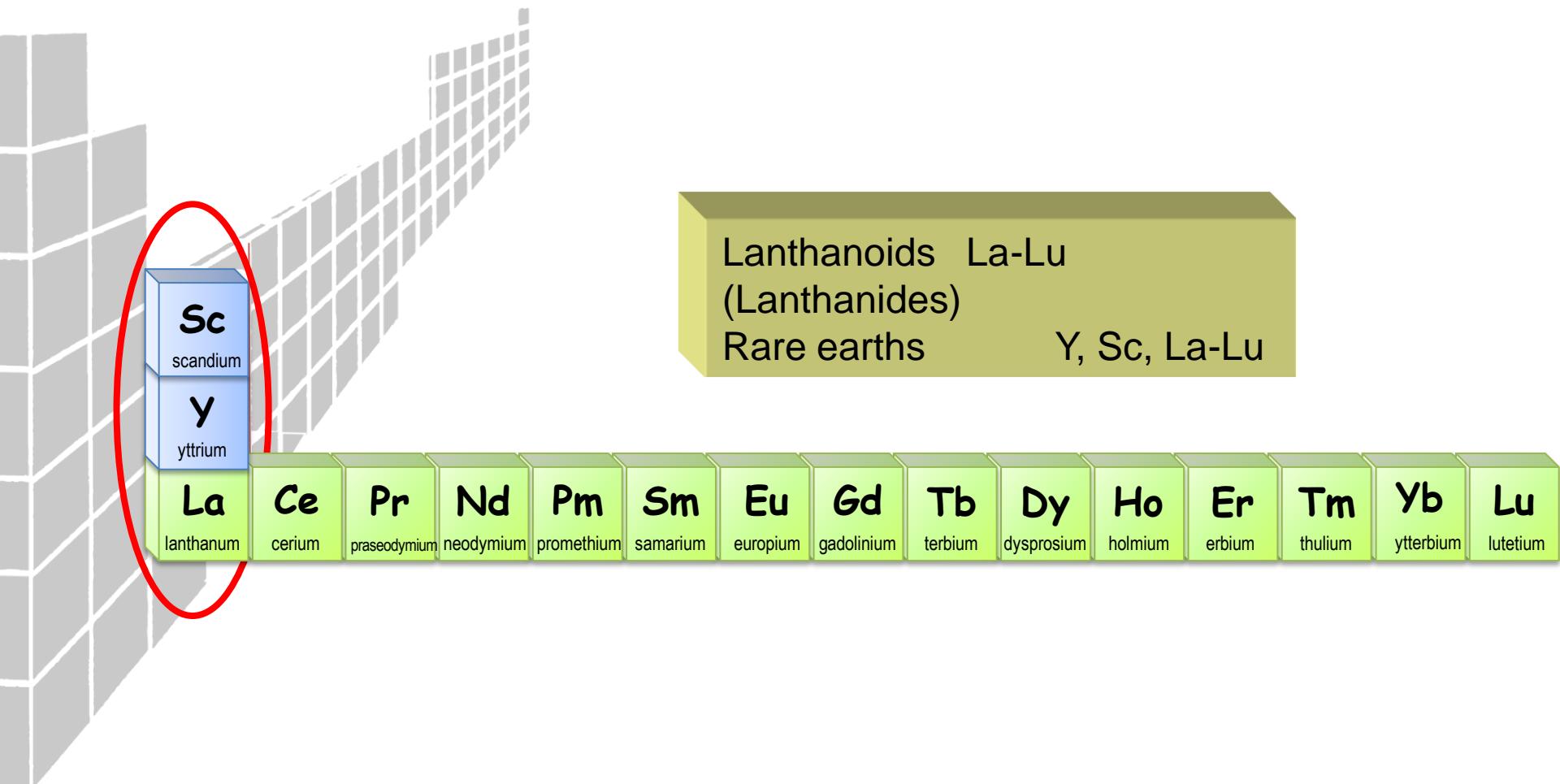
G. Merker, A. Kamptner, P. Schmidt, UVRFIA, Nov. 2012

R. Gerhard Merker  
- MinERS-pm



# 1. The Rare Earth Elements (REE)

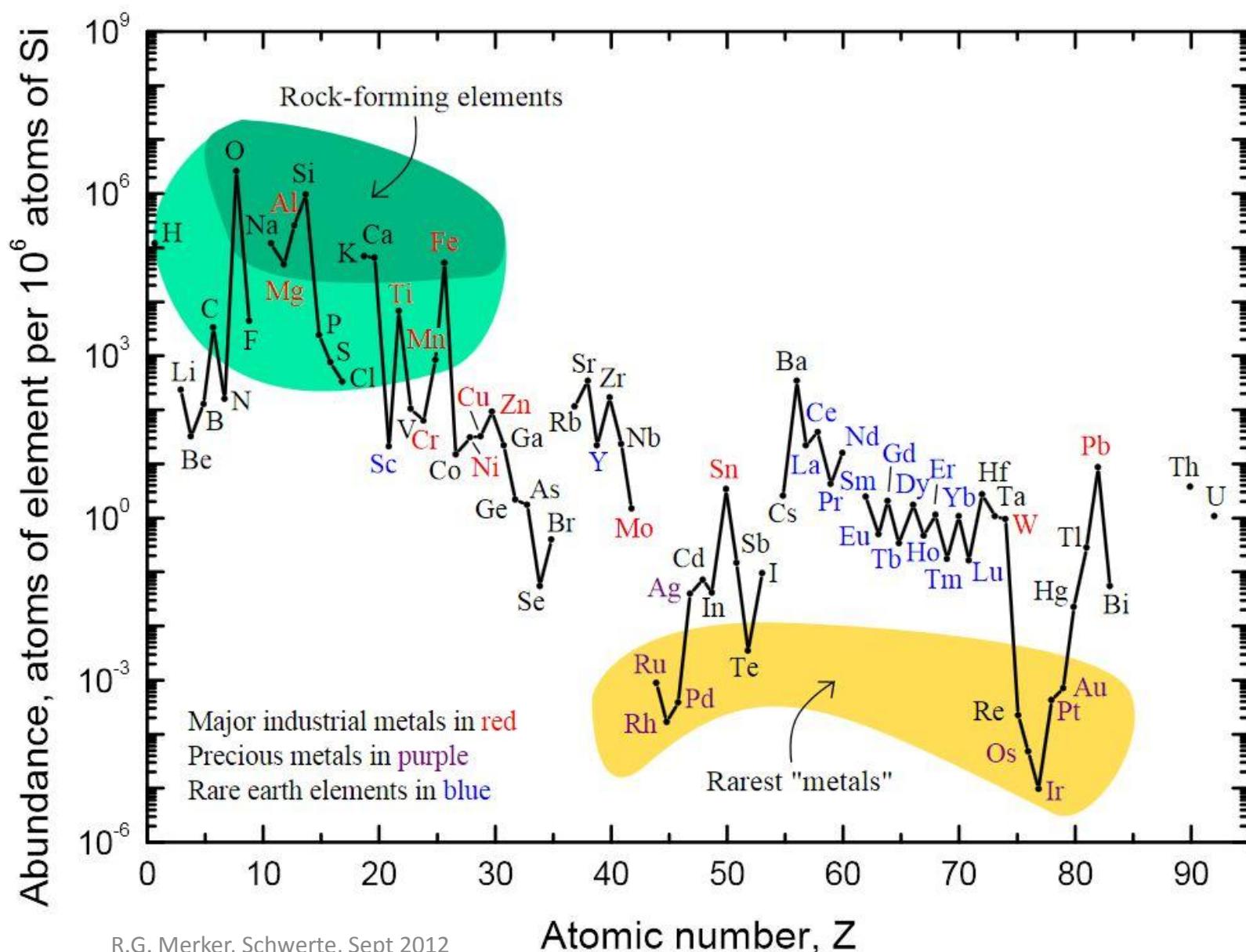
(courtesy of: Prof. J.-C. G. Bünzli, EPFL Lausanne/CH, /15/)



- "Ytter earth" (Y+Er+Tb-Oxide): extracted in 1794 by Johan Gadolin from the mineral Gadolinite which was discovered 1787
- Berzelius + Klaproth discovered in 1803 the so-called "Cerit earth" (La+Ce+Didym-Oxide) in the mineral Cerite that was found in 1751
- Mosander et al. cracked the "earthes" and separated their components from 1839 on

## 2. Are the REE really rare?

Sources: /15/: Bünzli REEC Münster,  
/20/: wikipedia.org/wiki/File:Elemental\_abundances.svg



### 3.1. Natural occurrence of REE

sources: various web sites /31/, (see sources-slide)

Phosphate



F-Carbonate



Oxide



Silicate



## 3.2. Natural occurrence of REE

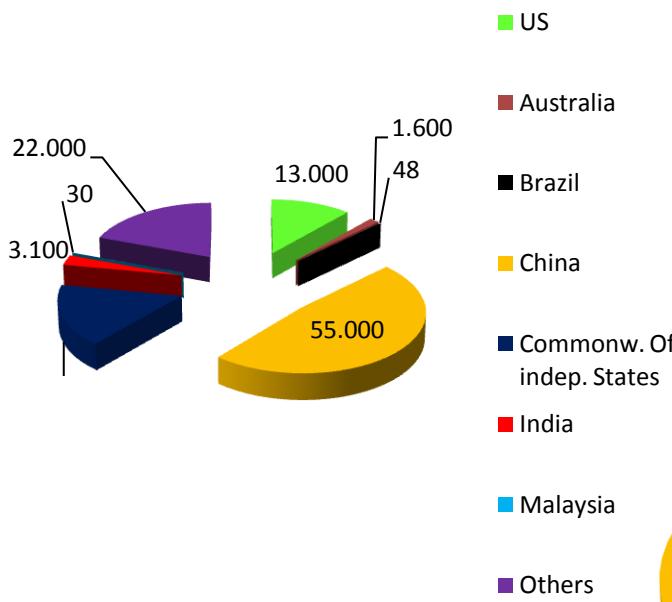
**sources:** Bulatovic /1/ corrected, Roesler /6/, Öko-Institut /7/, Mineralienatlas /9/, H. Richter /18/

Mineral	Simplified formula	Assay and REO content	true density	Mohs hardness
Monazite	(Ce,La... Th)PO <sub>4</sub>	50-68% RE <sub>2</sub> O <sub>3</sub>	4.9	5.5
Bastnaesite	(Ce,La,...)[CO <sub>3</sub> /F]	36-56% Ce <sub>2</sub> O <sub>3</sub> , ard. 19% La <sub>2</sub> O <sub>3</sub> ,	4.5-5.2	4-4.5
Synchisite	Ca (Ce, La, ...)[(CO <sub>3</sub> ) <sub>2</sub> /F]		3.9-4.15	4.5
Xenotime	YPO <sub>4</sub>	52-62% Y <sub>2</sub> O <sub>3</sub> , +Th, ~5% U,	4.6	4.5
Parisite	Ca(Ce, La...) <sub>2</sub> [CO <sub>3</sub> ] <sub>3</sub> F <sub>2</sub>	53- 61% RE <sub>2</sub> O <sub>3</sub>	4.3	4.5
Loparite	(Na,Ca,Ce,Sr) <sub>2</sub> Ti,Ta,Nb) <sub>2</sub> O <sub>6</sub>	42-45 % RE <sub>2</sub> O <sub>3</sub> , 8-11% (Ta,Nb) <sub>2</sub> O <sub>5</sub> , +Th	4.8	6
Fergusonite	(Y,Sr,Ce,U)(Nb,Ta,Ti)O <sub>4</sub>	31-42% Y <sub>2</sub> O <sub>3</sub> , 1-4% ThO <sub>2</sub> , 1-6% UO <sub>2</sub>	5.6-6.2	5.5-6.5
Eudyalite	Na <sub>3</sub> (Ca,Fe, Ce...) <sub>3</sub> Zr((OH,Cl)/(Si <sub>3</sub> O <sub>9</sub> ) <sub>2</sub> )	up to 2.9 % REO	2.8-3	5-5.5
Aeschynite	(Ce,Ca,Th,Y)(Ti,Nb) <sub>2</sub> O <sub>6</sub>	20-24% RE <sub>2</sub> O <sub>3</sub> , 11-17% ThO <sub>2</sub>	4.9-5.1	5-6
Lateritic clays	Adsorbed ions	Ce <sub>2</sub> O <sub>3</sub> 1-7 %, Nd <sub>2</sub> O <sub>3</sub> 5-30, Y <sub>2</sub> O <sub>3</sub> 10-63 %	-	-

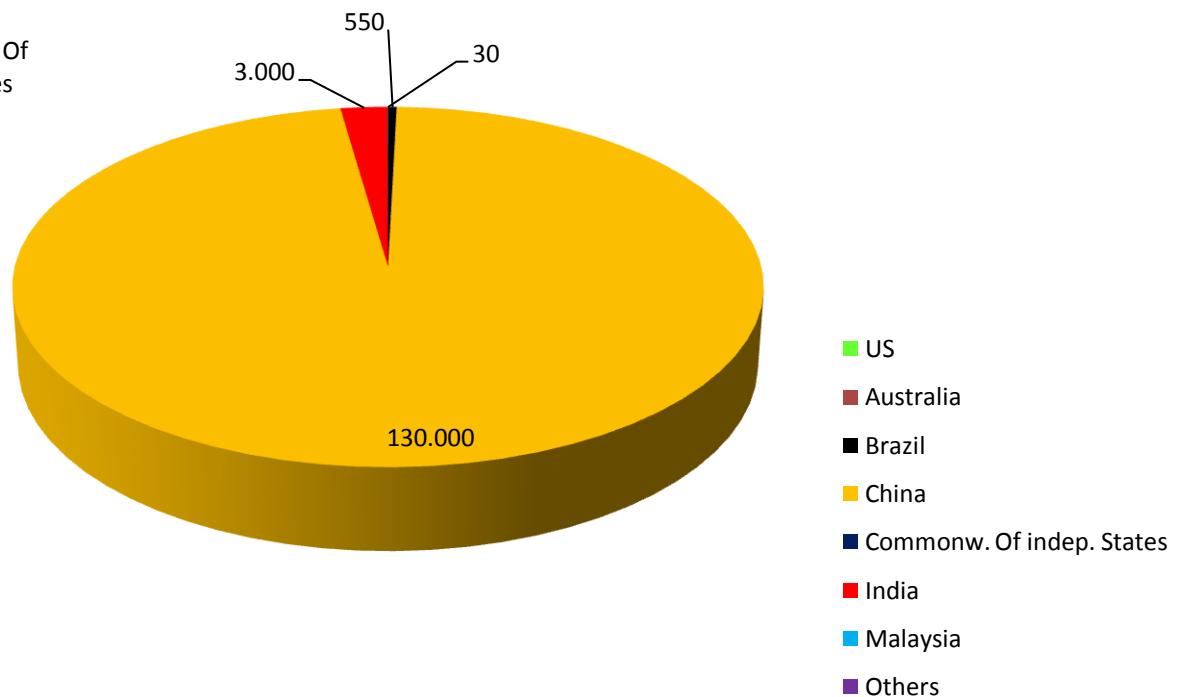
## 4. Who produces RE?

source /19/: USGS RE fact sheet 2012

RE reserves 2012 (x 1.000t)



RE mine production 2011 mt/a



# 5. Bayan Obo / Baotou: Capital of REE

Source /16/, Bünzli REEC

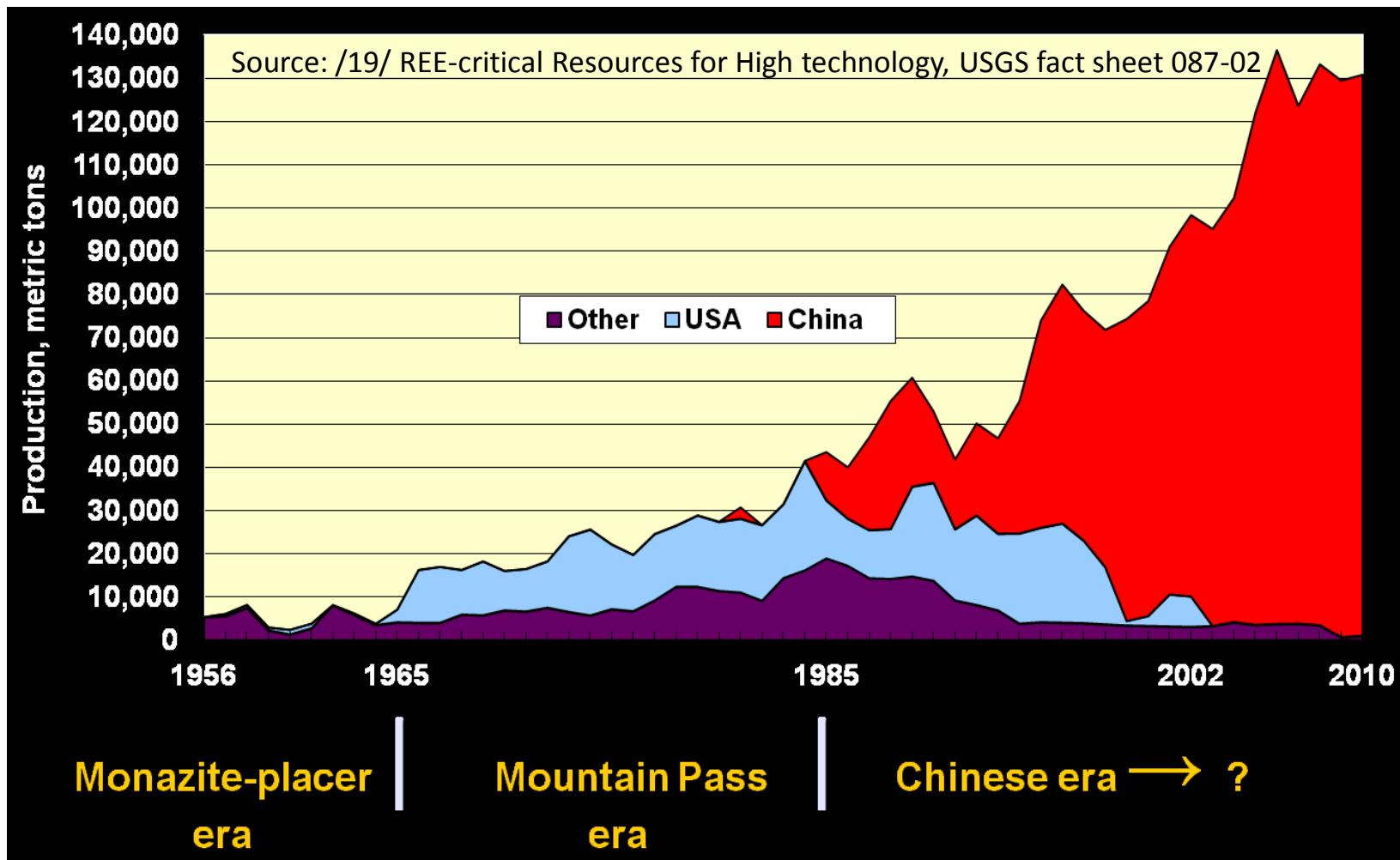


**Baotou iron mining / Inner Mongolia –  
Source of nearly 60 % of worlds TREO  
production  
(estimation based on /7/)**



## 6. Why about Carbonatites?

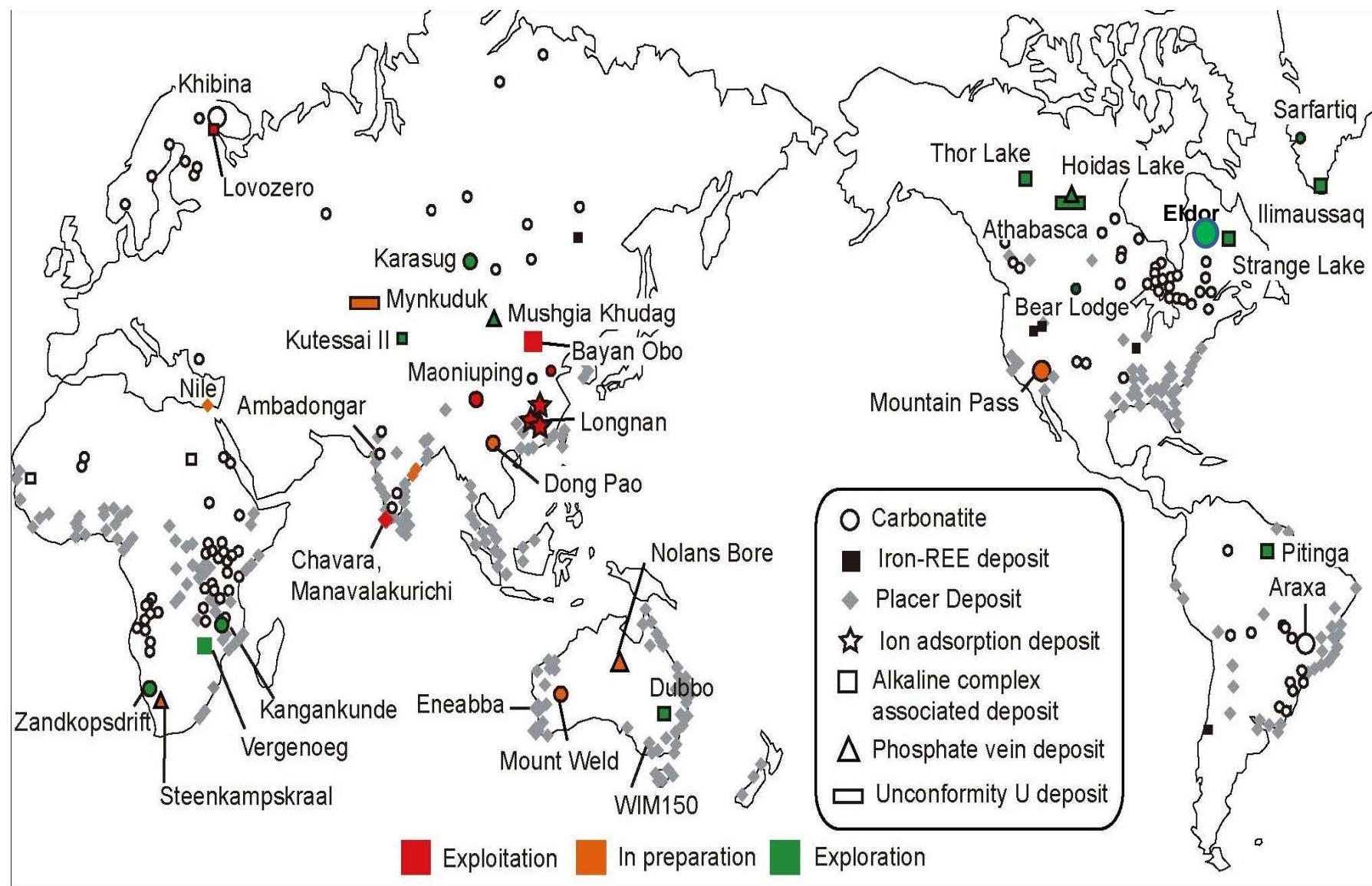
from placer to carbonatite deposits



Global rare earth element production ( $1 \text{ kt} = 10^6 \text{ kg}$ ) from 1950 through 2000, in four categories: United States, almost entirely from Mountain Pass, California; China, from several deposits; all other countries combined, largely from monazite-bearing placers; and global total. Four periods of production are evident: the monazite-placer era, starting in the late 1800s and ending abruptly in 1964; the Mountain Pass era = start of the carbonatite era,

# 7. Selected RE-Projects, (out of some 400-550)

(source /16/: [www.matamec](http://www.matamec));



### 3. Selected natural factors of the processing of k-RE-ores

Factor	RE-ores
Paragenesis (Association)	Chemically similar associated minerals, same Cat- and Anions (Ca, F, CO <sub>3</sub> ), ... tricky separation by flotation
Weathering	Slime formation trend can occur, even enrichment of RE within the slime fraction
Grain size / Locking	Often finely intergrown, partly multi-stage milling down below 20 microns needed
Milling properties / hardness	Enrichment in the fines can occur
Geochemical spread of the RE	Usually bound by various RE-Minerals, even partly dissemination into the lattice of gangue minerals possible (limited recovery!)
Magnetic properties	fluctuating, partly weak-magnetic properties, partly not
Density	Usually pretty high, but often too fine grain size for a gravity separation
Flotability / surface properties	Often similar by chemical alikeness of the associated minerals, causes usually more or less complicated multi-stage techniques (several collectors and depressants, pH-control, water-exchange, desorption, hot processing, a. o. gadgets.)

**Each and every ore / RE-project shows different features and edge conditions.  
Jedes Erz / jedes Projekt hat eine andere Charakteristik und zeigt unterschiedliche Randbedingungen!!!**

## 9. The classic carbonatite project: Mountain Pass/US

source /11/, Molycorp website 2012



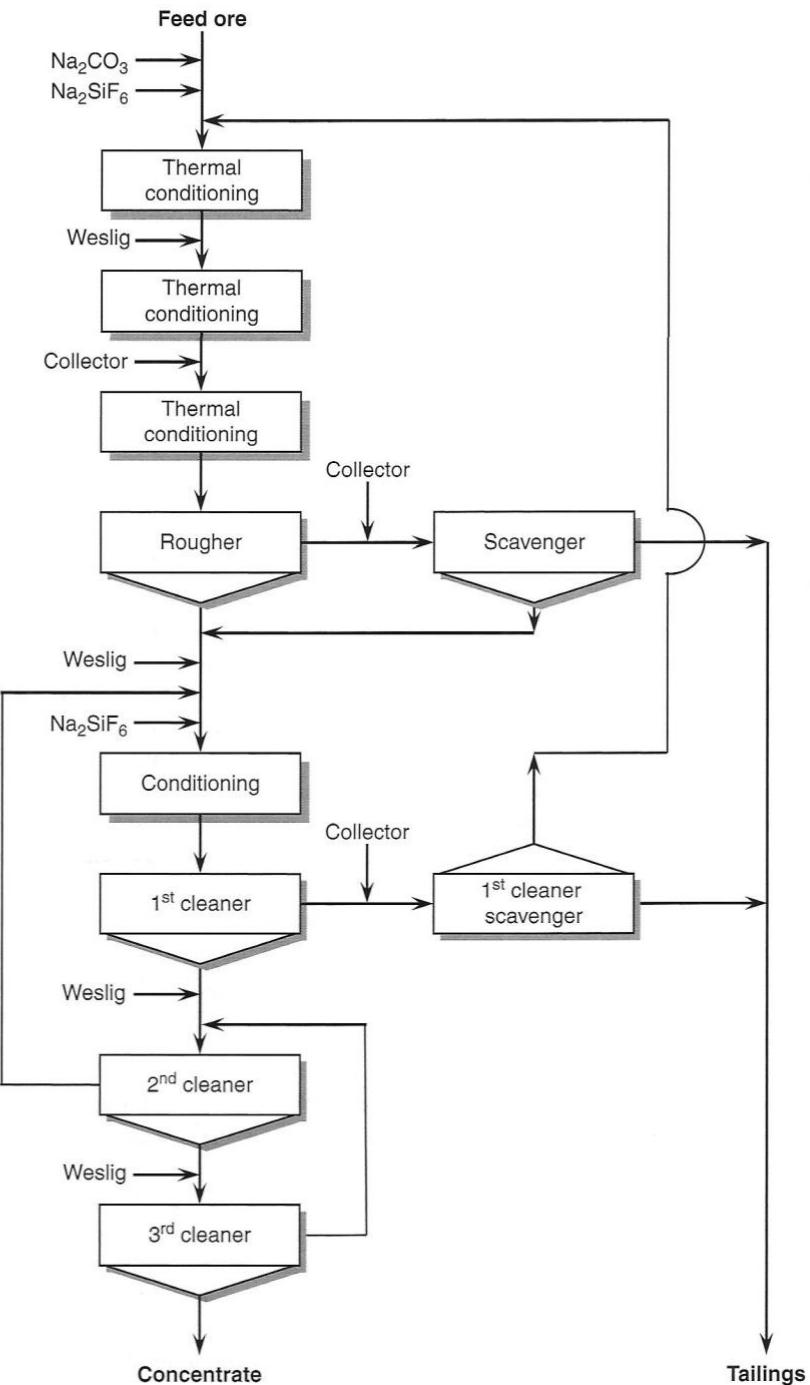
Mountain Pass overview and processing facility,



# 10. Processing of carbonatite RE-ores, Flow sheet Mountain Pass,

Source: /1/ Bulatovic

Product	Weight (%)	Assays (%)					% Distribution	
		REO	Ce <sub>2</sub> O <sub>3</sub>	La <sub>2</sub> O <sub>3</sub>	BaSO <sub>4</sub>	CaO	REO	Ce <sub>2</sub> O <sub>3</sub>
Final bastnaesite concentrate	9.38	64.1	31.4	22.2	2.7	3.1	75.6	75.5
Final bastnaesite tailing	90.62	2.28	1.06	0.74	26.3	16.9	24.4	24.5
Feed	100.00	8.09	3.9	2.76	26.3	15.6	100.0	100.0



# 11. The presumably next big project, Mount weld / Au, pit and concentration plant

Source: /12/ web site of Lynas corp. 2012



## 12. Processing of RE-Carbonatite ores, proposed flow sheet Mount Weld

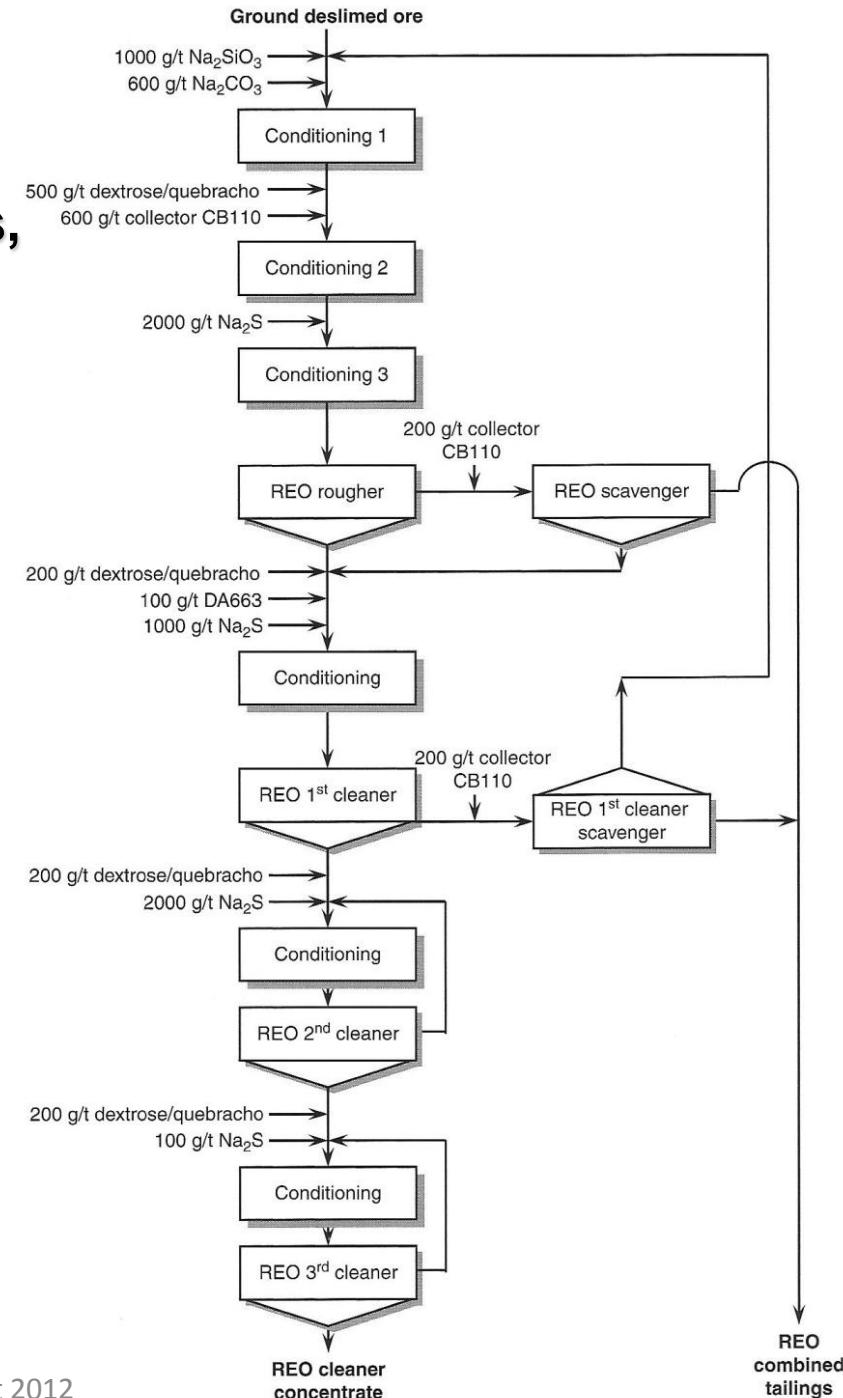
Source /1/ Bulatovic 2010

Head analyses of the Mount Weld ore

Element	Assays (%)
Total REO	15.50
Cerium ( $\text{Ce}_2\text{O}_3$ )	9.54
Lanthanum ( $\text{La}_2\text{O}_3$ )	4.21
Samarium ( $\text{Sm}_2\text{O}_3$ )	0.39
Yttrium ( $\text{Y}_2\text{O}_3$ )	0.30
Iron ( $\text{Fe}_2\text{O}_3$ )	60.5
Alumina ( $\text{Al}_2\text{O}_3$ )	15.5
Magnesia ( $\text{MgO}$ )	4.60
Calcium ( $\text{CaO}$ )	10.8
Phosphorus ( $\text{P}_2\text{O}_5$ )	2.66

Overall metallurgical results obtained on the Mount Weld ore

Product	Weight (%)	% Monazite assay	% Monazite recovery
Cleaner concentrate	20.89	58.5	77.5
Combined tail	54.51	2.55	8.8
Slimes	24.6	8.8	13.7
Feed	100.00	15.8	100.0



## 13. RE-carbonatite investigations in Germany?

- 1970ies and 80ies: Storkwitz carbonatite, ZGI Berlin, IfR Dresden, FIA Freiberg, exploration core drillings and processing tests; (currently DRAG)
- Beginning of the 1980ies: Bayan Obo ore, KHD Humboldt Wedag Cologne, development of a processing technique for the Fe-RE-ores
- 1980ies: geological exploration and process development, Feasibility Study Dong Pao/Vietnam, IfR Dresden + FIA Freiberg
- 2012: Participation in processing tests of a Feasibility Study of an international RE-project (Eldor), UVR-FIA + consultant
- more projects in the pipeline?

# 14. Project Dong Pao, pictures of the RE-Deposit of Dong Pao

Source /22/: R.G.Merker



R.G. Merker, Schwerte, Sept 2012

# 15. Aufbereitung von RE-Karbonatiterzen, weathered ore of Dong Pao

Source /2/: Merker et al.: Erzmetall 1991 etc.

Table 2 – Structure of grain sizes, together with distribution of RE depending upon grain size, as demonstrated by a mixed sample of ore type I

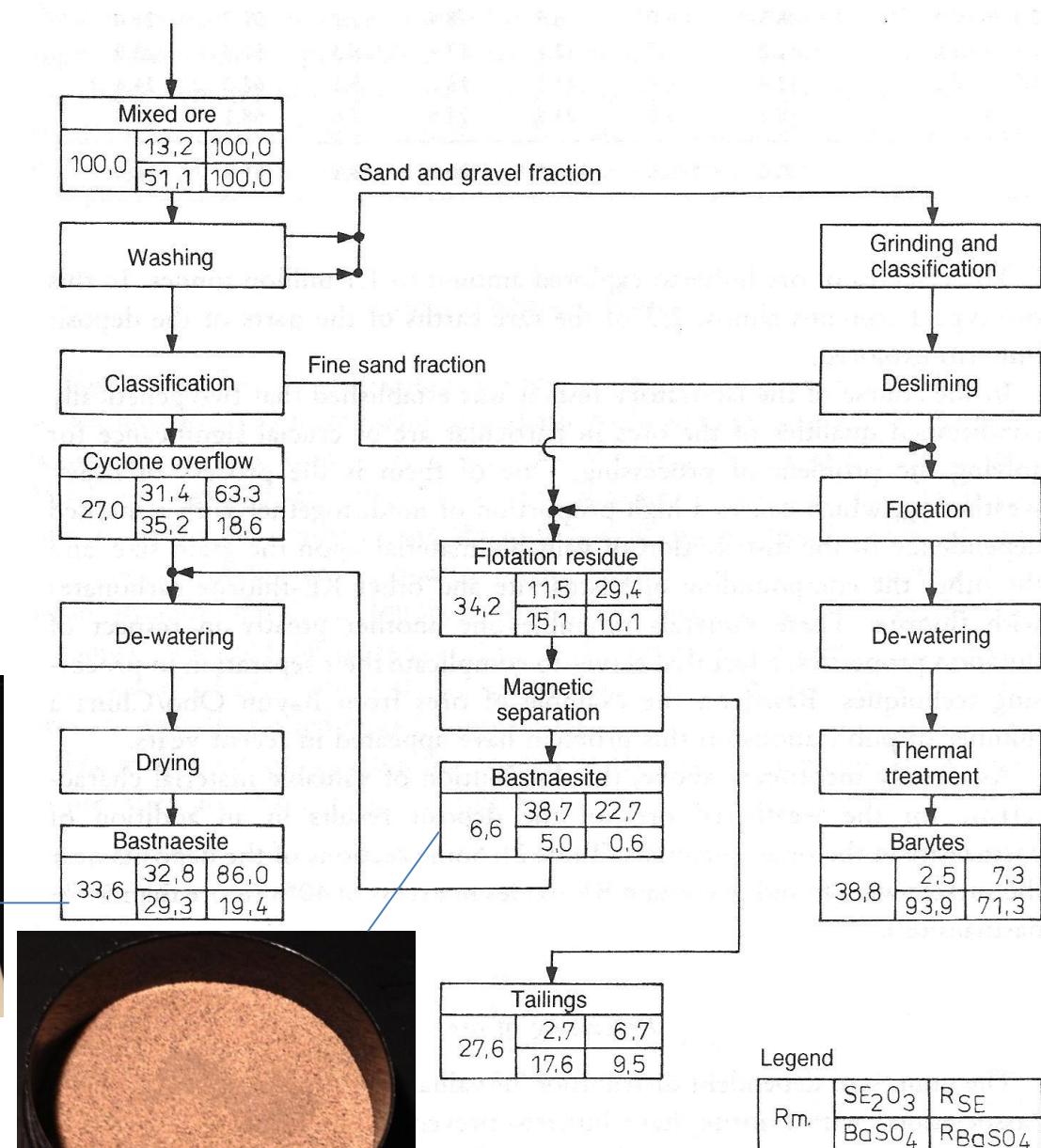
Grain size in mm	Distribution in %				Assay in %		
	Mass	SE <sub>2</sub> O <sub>3</sub>	Barytes	Fluorite	SE <sub>2</sub> O <sub>3</sub>	Barytes	Fluorite
< 0.015	16.8	43.9	7.5	4.1	35.7	22.3	3.9
0.015 – 0.032	7.4	8.0	7.3	5.1	14.7	48.8	11.2
0.032 – 0.063	5.7	5.4	5.3	4.3	12.5	48.6	12.7
0.063 – 0.1	9.1	9.7	10.2	7.3	13.9	58.3	13.4
0.1 – 0.315	11.1	10.9	11.4	10.9	12.9	53.4	16.5
0.315 – 0.5	6.5	6.0	6.5	8.4	12.1	51.7	21.6
0.5 – 1.0	12.2	7.7	12.2	17.4	8.3	51.9	23.9
1.0 – 2.0	12.4	4.8	14.8	18.0	5.1	62.0	24.4
> 2.0	18.8	3.6	24.8	24.5	2.6	68.1	21.8
	100.0	100.0	100.0	100.0	13.1	51.9	16.8



# 16. Processing of RE-Carbonatite ores, one of the proposed flow sheets

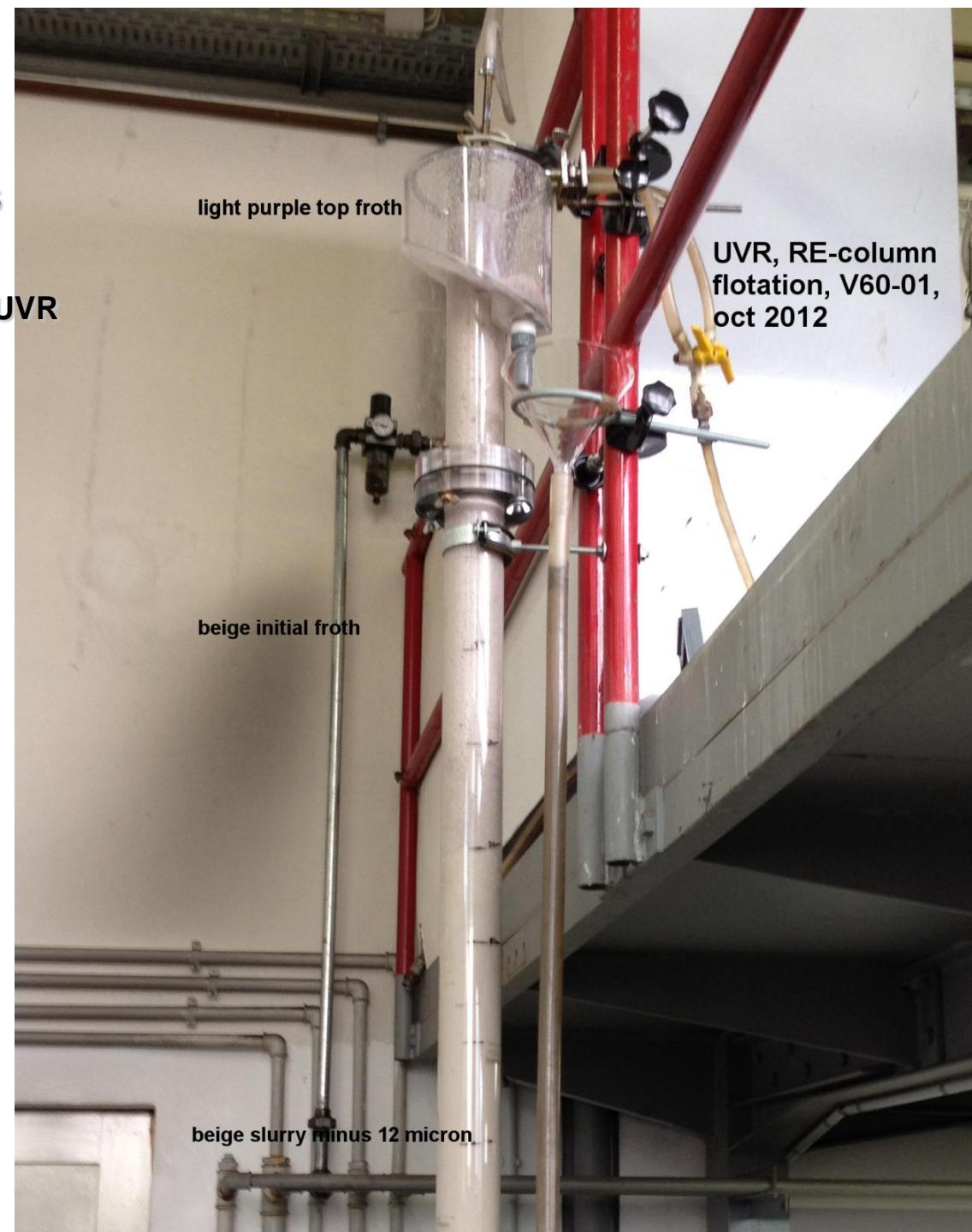
Dong Pao  
(IfR+FIA 1990)

Source /2/: Merker et al.: Erzmetall 1991



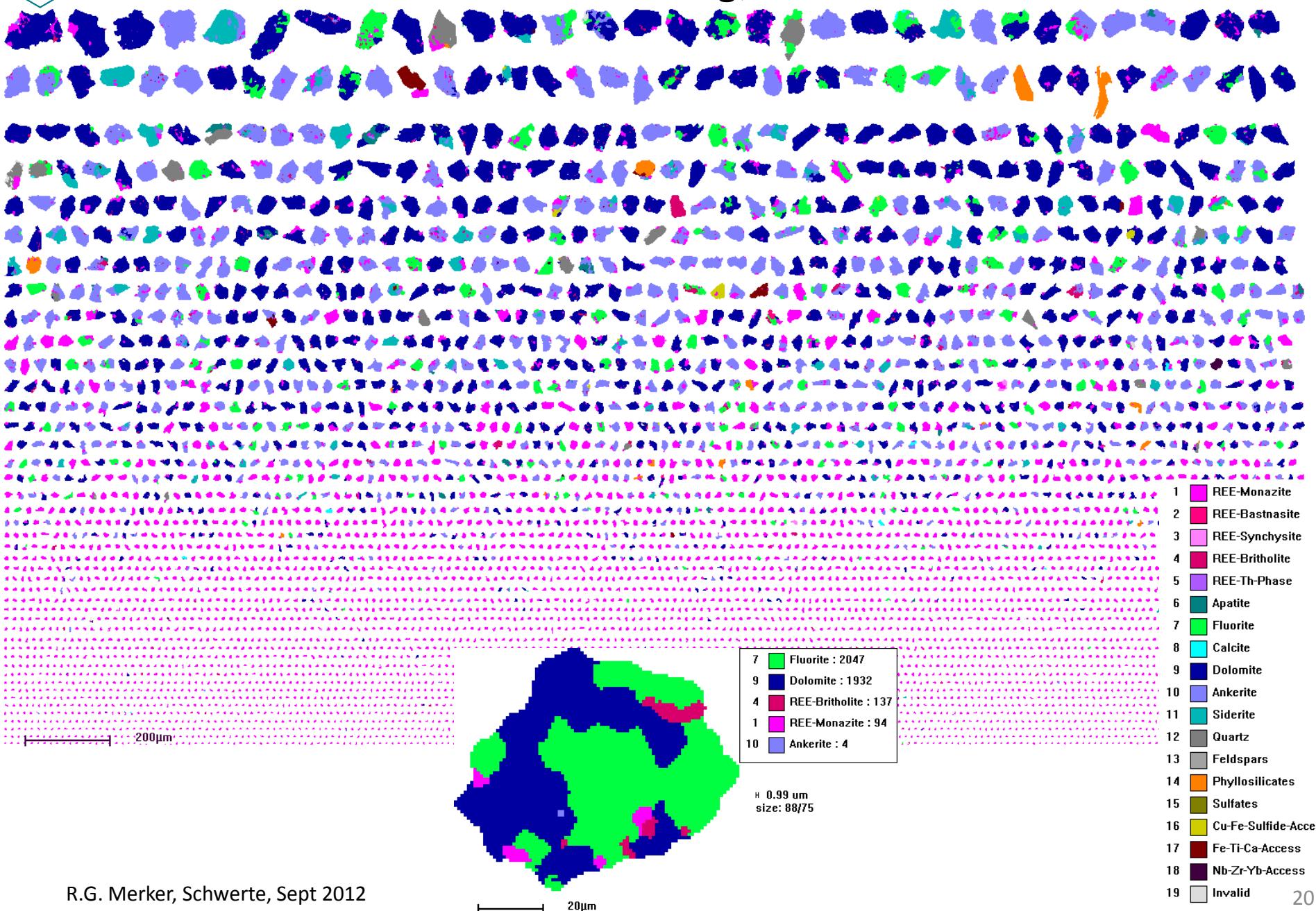
for processing Dong Pao Rare Earth ores. Commentary in text.

## 17. Current RE-projects of UVR-FIA, column flotation, source /15/ UVR

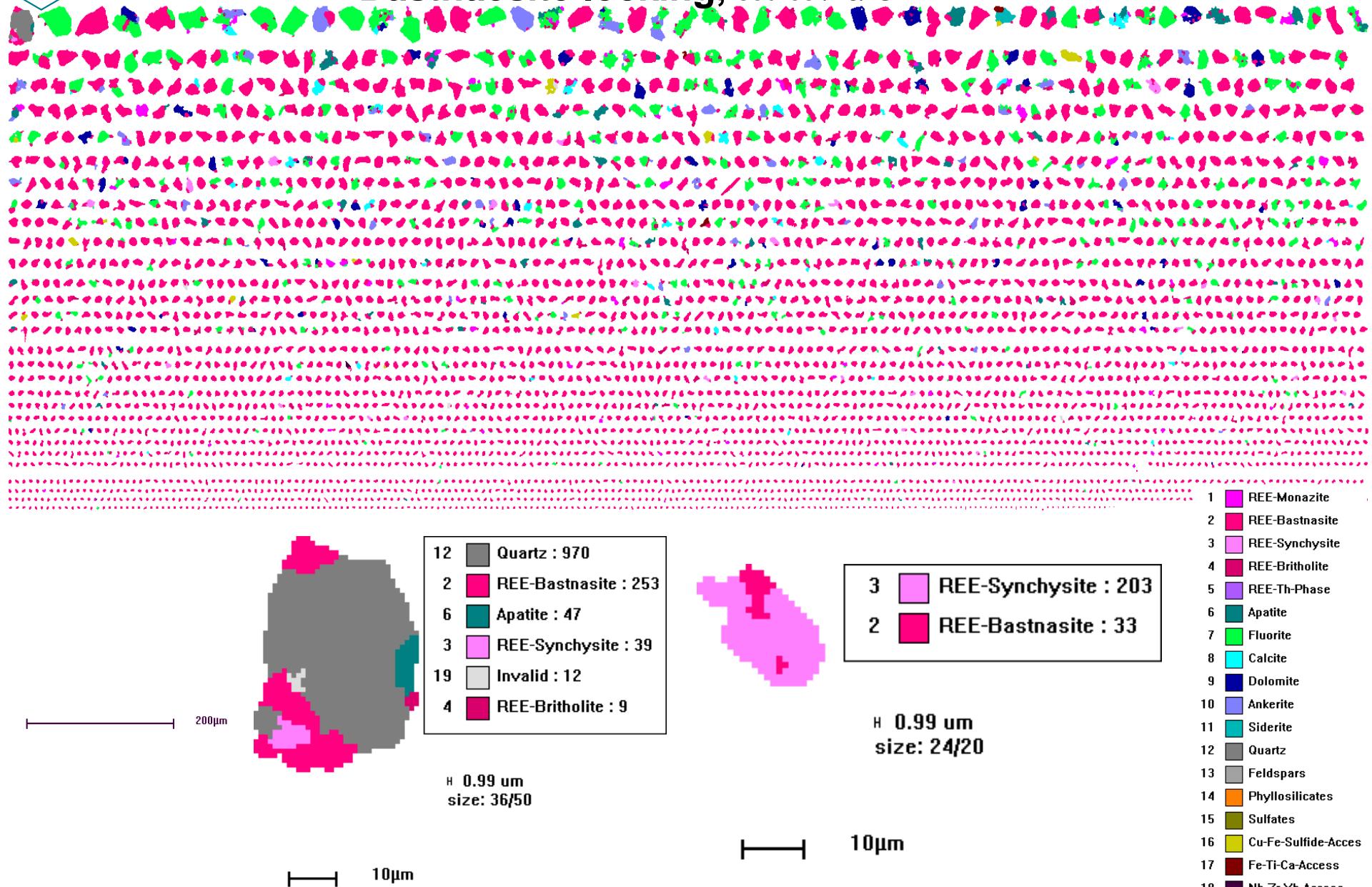


# 18.1. UVR project E/A, Ore III 45, minus 70 µ

## Monazite locking, source /15/ UVR-FIA

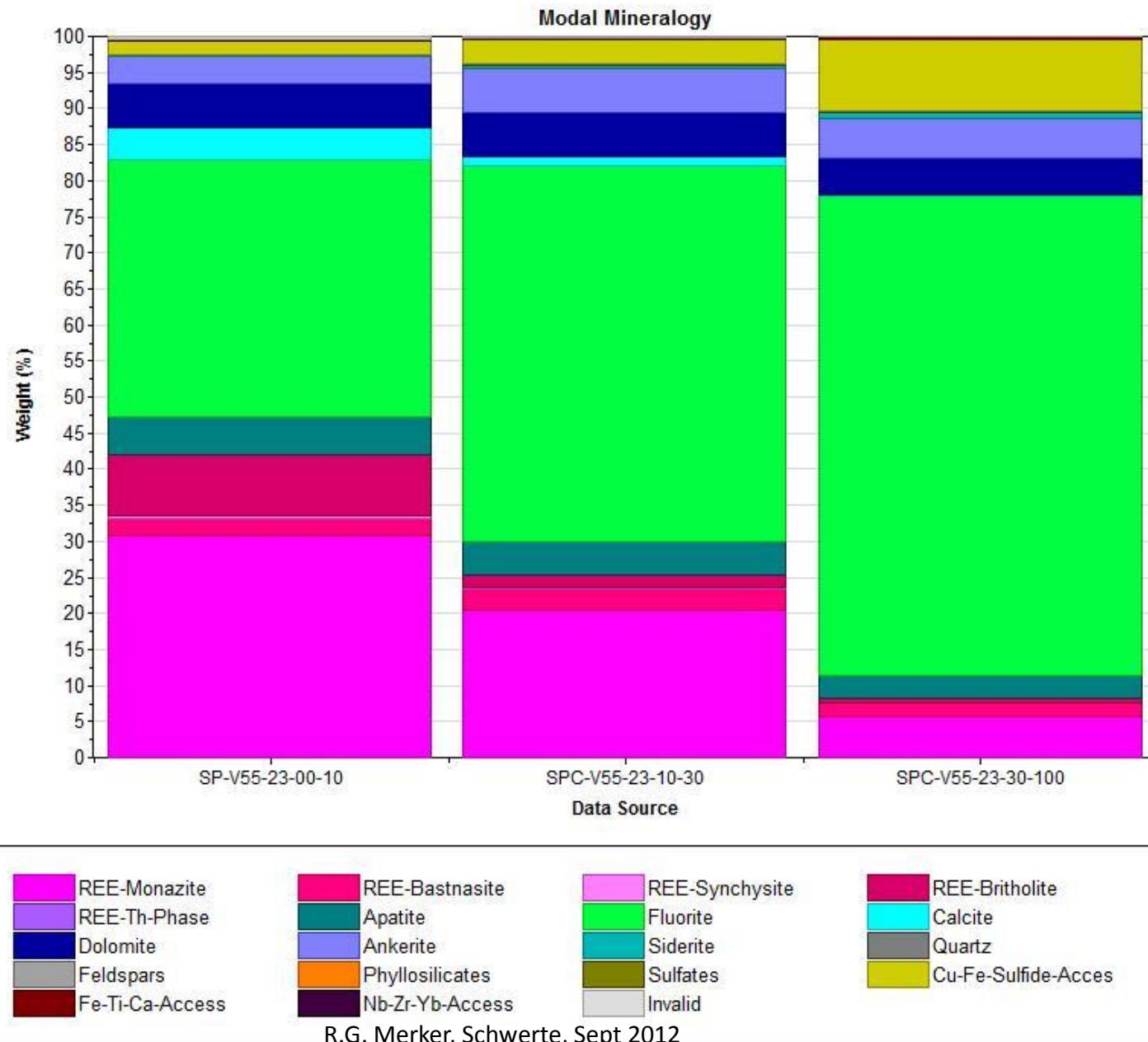


# 18.2. UVR project E/A, flot. Con 55-23, Bastnaesite locking, source /15/ UVR-FIA



# 19. Mineralogy of bulk flotation, con UVR-E/A

MLA TU BAF Freiberg, source /15/ UVR-FIA



## 20. Selected technical factors of the Flotation of RE-minerals

Faktor	Faktor
Natürlicher pH	Gelöste Ionen / Wasserqualität
pH Regler (NaOH, HCl, H <sub>2</sub> SO <sub>4</sub> etc.)	Temperatur (Bsp. MP, Storkwitz und XA)
Drücker (Bsp XA Lignex / Ultrazine)	Flotationszeit
Sammlertyp / chem. Modifikationen / Kombinationen	Konditionierung / Konditionierungszeit
Aktivatoren	Schaumstabilität / Schäumer
Selektivität der Reagenz-Kombination	Persönliche Einflussfaktoren
Reagenzmengen	Dispergierung / Turbulenz
Desorbierbarkeit	Feststoffgehalte Trübe
Mahlung / Mahlfeinheit	Geometrische Faktoren (Größe, Form der Zellen etc.)
Schlammbildung / Entschlämmung	Luftmenge / Blasengröße

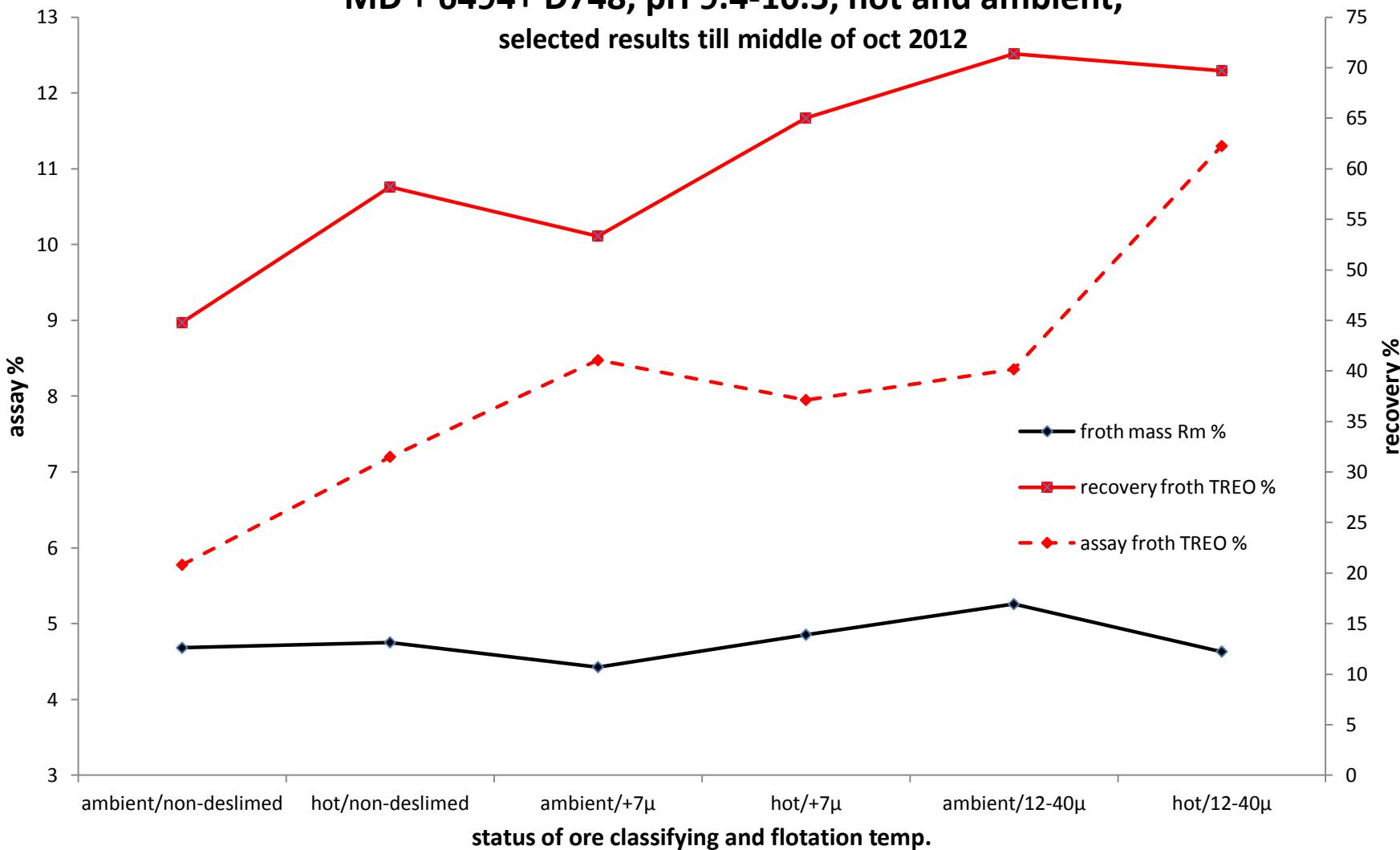
# 21.1. Selected details about own results, influence of grain size and T on flotation, E/A

Source /15/, UVR-FIA

rougher flotation of ore III and IV, dependency on classifying, standard

MD + 6494+ D748, pH 9.4-10.5, hot and ambient,

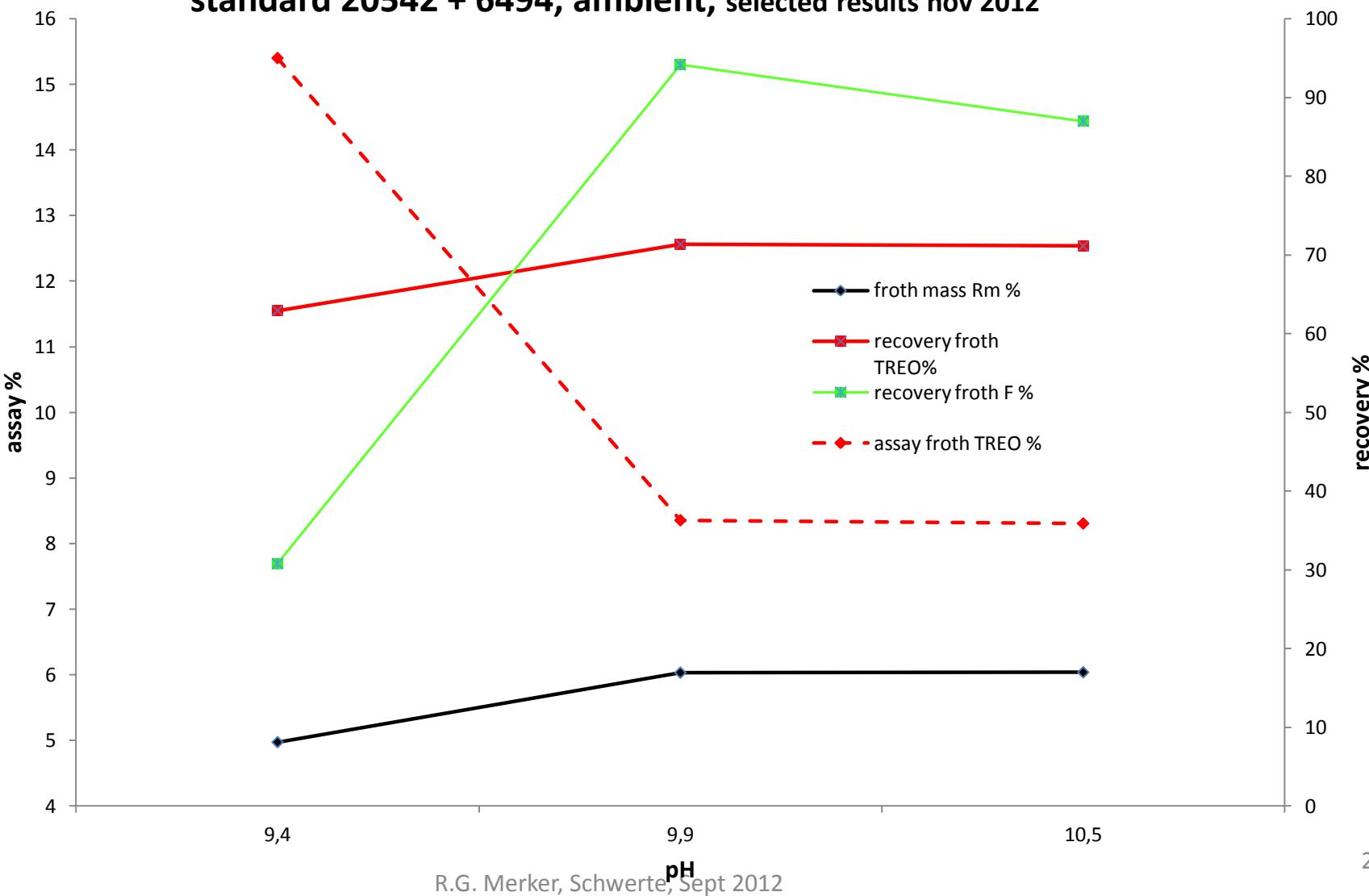
selected results till middle of oct 2012



## 21.2. Selected details about own results, pH influence, UVR-E/A

Source /15/, UVR-FIA

rougher flotation ore IV, 12-40 microns, pH dependency,  
standard 20542 + 6494, ambient, selected results nov 2012



## 22. Current RE-projects of UVR, fancy colors by flotation, source /15/ UVR



R.G. Merker, Schwerte, Sept 2012

## 23. Thank you for your kind attention!

**UVR-FIA can help pushing your RE-Project too!  
Don't be shy – contact us!**

**Ihr Anlaufpunkt in Fragen der  
RE-Erzaufbereitung: UVR-FIA + G.Merker**

**R. Gerhard Merker - Mineral Processing Engineer**  
(Mineral engineering, Recycling, Sourcing, Consulting,  
Project Management)

58239 Schwerte / Germany  
Brunsiepen 11  
Tel: +49 2304 331 750  
Cell: private +49 1577 1979 144  
Cell: business. +49 173 5355 828  
skype: r.gerhard.merker

E-mail: [merker@merker-mineral-processing.de](mailto:merker@merker-mineral-processing.de)  
web-site: [www.merker-mineral-processing.de](http://www.merker-mineral-processing.de)



# 24. Sources

- (1) S. M. Bulatovic: handbook of Flotation Reagents, Vol 2, Elsevier BV. 2010
- (2) G. Merker et al.: Geol. Situation and Proc. Possibilities of the RE-Ore of Dong Pao /Vietnam, in Erzmetall 44 (1991), Nr. 9, p. 452 ff.
- (3) Final report of Invest. IfR Dresden 1984: G. Merker u. U. Schramm: .... Investigation of the carbonatite of Storkwitz
- (4) G. Lasch et. Al.: Rohstoffforschung am IfR Dresden ...., publication series Geowissenschaften 19, Verlag Störr 2012
- (5) Website [www.Australianrareearths.com](http://www.Australianrareearths.com)
- (6) H. J. Rösler: School book of Mineralogy, Dt. Verlag f. Grundstoffindustrie 1981
- (7) Öko-Institut Freiburg; Study of RE and their recycling 2012
- (8) D. J. Kingsnorth: Rare Earths ,Supply Security – Dream or Possibility, Konferenz Strategische Rohstoffe-Risikovorsorge, Freiberg April19/20 2012
- (9) Website, [www.Mineralienatlas.de](http://www.Mineralienatlas.de), sept 2012
- (10) Website, [www.Techmetalsresearch.com](http://www.Techmetalsresearch.com), sept. 2012
- (11) Website of Molycorp Mountain Pass, [www.molycorp.com](http://www.molycorp.com), sept 2012
- (12) Website of Lynas Corp. Ltd, [www.lynascorp.com](http://www.lynascorp.com), sept 2012
- (13) Report of Investigation ZGI Berlin 1981: G. Röllig, N. Reuter: State report and analysis Carbonatite Delitzsch
- (14) G. Merker et. Al.: Geological structure and processing poss. of the rare earth ore of Dong Pao, in Nat. res. and developm. Vol. 37, 1993, p. 7ff, Inst. Of Scientif. Co-op. Tübingen
- (15) Internal information from UVR-FIA about currant RE-Projects
- (16) Prof. J.-C. G. Bünzli, EPFL Lausanne: The fascinating world of lanthanide luminescence, REEC paper Münster Sept 2012
- (17) Website [www.matamec.com](http://www.matamec.com), sept 2012
- (18) H. Richter: Specialized paper: The ind. Prod. of RE in China, in Erzmetall 49 (1996) Nr. 2, S. 134ff
- (19) L. Andresen, P. Zahr: Processing possibilities of the complex Iron ores of Baotou, Erzmetall 39 (1986), Nr. 5, S.240 ff
- (20) USGS website, RE fact sheet 2012 u.a.
- (21) web site, [http://en.wikipedia.org/wiki/File:Elemental\\_abundances.svg](http://en.wikipedia.org/wiki/File:Elemental_abundances.svg)
- (22) R.G. Merker: private pictures
- (23) L. Lesch: internal calculation of the RE deposit Dong Pao/Vietnam, IfR Dresden 1987
- (24) Luo Jiake et. al.: A new dev. of mineral proc. flow sheet for the treatment of a complex ore containing RE, F, Nb, Fe; XVth int. Min. Proc. Congr., June 1985, Cannes, Bd. III p. 474 ff.
- (25) Web site [www.alkane.com](http://www.alkane.com) , sept 2012
- (26) Web site, [www.avalonraremetals.com](http://www.avalonraremetals.com), sept 2012
- (27) Web site [www.stansenergy.com](http://www.stansenergy.com), sept 2012
- (28) Web site [www.ggg.gl](http://www.ggg.gl), sept 2012
- (29) Web site [www.rareelementsresources.com](http://www.rareelementsresources.com), sept 2012
- (30) Website [www.rareelementsresources.com](http://www.rareelementsresources.com), pdf technical report Roche engineering
- (31) Various web sites:<http://i.ebayimg.com>; <http://2.bp.blogspot.com>; <http://www.irocks.com/>